

## 5-Fluoro-1*H*-indole-3-carboxylic acid

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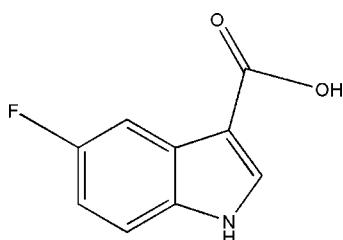
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Key indicators: single-crystal X-ray study;  $T = 293\text{ K}$ ; mean  $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$ ;  
 $R$  factor = 0.054;  $wR$  factor = 0.137; data-to-parameter ratio = 14.5.

In the title compound,  $\text{C}_9\text{H}_6\text{FNO}_2$ , the carboxyl group is twisted slightly away from the indole-ring plane [dihedral angle =  $7.39(10)^\circ$ ]. In the crystal, carboxyl inversion dimers linked by pairs of  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds generate  $R_2^2(8)$  loops and  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds connect the dimers into  $(10\bar{1})$  sheets.

### Related literature

For background to indoles as pharmaceuticals, see: Lang *et al.* (2011).



### Experimental

#### Crystal data

$\text{C}_9\text{H}_6\text{FNO}_2$   
 $M_r = 179.15$

Monoclinic,  $P2_1/n$   
 $a = 4.4176(9)\text{ \AA}$

$b = 11.073(2)\text{ \AA}$   
 $c = 16.014(3)\text{ \AA}$   
 $\beta = 96.63(3)^\circ$   
 $V = 778.1(3)\text{ \AA}^3$   
 $Z = 4$

Mo  $K\alpha$  radiation  
 $\mu = 0.13\text{ mm}^{-1}$   
 $T = 293\text{ K}$   
 $0.30 \times 0.20 \times 0.10\text{ mm}$

#### Data collection

Rigaku SCXmini CCD diffractometer  
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2005)  
 $T_{\min} = 0.982$ ,  $T_{\max} = 0.993$

7874 measured reflections  
1788 independent reflections  
1153 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.068$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.054$   
 $wR(F^2) = 0.137$   
 $S = 1.02$   
1788 reflections  
123 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\max} = 0.18\text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.26\text{ e \AA}^{-3}$

**Table 1**  
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

| $D-\text{H}\cdots A$             | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|----------------------------------|--------------|--------------------|-------------|----------------------|
| O1—H1 $\cdots$ O2 <sup>i</sup>   | 0.82         | 1.86               | 2.669 (2)   | 171                  |
| N1—H1B $\cdots$ O2 <sup>ii</sup> | 0.89 (3)     | 2.18 (3)           | 3.026 (2)   | 159 (2)              |

Symmetry codes: (i)  $-x + 2, -y + 1, -z + 2$ ; (ii)  $-x + \frac{3}{2}, y - \frac{1}{2}, -z + \frac{3}{2}$ .

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *DIAMOND* (Brandenburg & Putz, 2005); software used to prepare material for publication: *SHELXL97*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6548).

### References

- Brandenburg, K. & Putz, H. (2005). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
- Lang, L., Wu, J.-L., Shi, L.-J., Xia, C.-G. & Li, F.-W. (2011). *Chem. Commun.* **47**, 12553–12555.
- Rigaku. (2005). *CrystalClear*. Rigaku Corporation, Tokyo, Japan.
- Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.

## **supplementary materials**

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### 5-Fluoro-1*H*-indole-3-carboxylic acid

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#### Experimental

A commercial sample of the title compound was obtained. Colourless prisms were obtained by slow evaporation of a methanol solution over a period of seven days.

#### Refinement

All H atoms attached to C atoms and O atoms were fixed geometrically and treated as riding with C—H = 0.93 Å (CH), N—H = 0.86 Å and O—H = 0.86 Å with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{CH})$ ,  $U_{\text{iso}}(\text{H}) = 1.3U_{\text{eq}}(\text{N})$  and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$ .

#### Figures

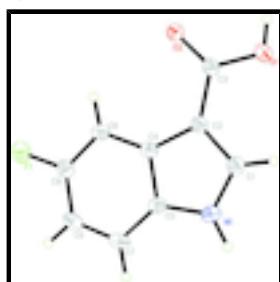


Fig. 1. The molecular structure of the title compound with displacement ellipsoids drawn at the 30% probability level.

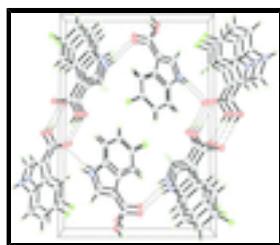


Fig. 2. A packing view down the  $a$  axis showing hydrogen bonds as dashed lines.

### 5-Fluoro-1*H*-indole-3-carboxylic acid

#### Crystal data

|  |   |
|--|---|
| C <sub>9</sub> H <sub>6</sub> FNO <sub>2</sub> | $F(000) = 368$  |
| $M_r = 179.15$                                 | $D_x = 1.529 \text{ Mg m}^{-3}$                         |
| Monoclinic, $P2_1/n$                           | Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$ |
| Hall symbol: -P 2yn                            | Cell parameters from 1954 reflections                   |
| $a = 4.4176 (9) \text{ \AA}$                   | $\theta = 3.2\text{--}25.0^\circ$                       |
| $b = 11.073 (2) \text{ \AA}$                   | $\mu = 0.13 \text{ mm}^{-1}$                            |
| $c = 16.014 (3) \text{ \AA}$                   | $T = 293 \text{ K}$                                     |

# supplementary materials

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$\beta = 96.63(3)^\circ$   
 $V = 778.1(3)\text{ \AA}^3$   
 $Z = 4$

## Data collection

Rigaku SCXmini CCD diffractometer  
Radiation source: fine-focus sealed tube graphite  
Detector resolution: 13.6612 pixels mm<sup>-1</sup>  
CCD profile-fitting scans  
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2005)  
 $T_{\min} = 0.982$ ,  $T_{\max} = 0.993$   
7874 measured reflections

Prism, colourless

$0.30 \times 0.20 \times 0.10\text{ mm}$

## Refinement

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.054$   
 $wR(F^2) = 0.137$   
 $S = 1.02$   
1788 reflections  
123 parameters  
0 restraints

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0575P)^2 + 0.142P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.18\text{ e \AA}^{-3}$

$\Delta\rho_{\min} = -0.26\text{ e \AA}^{-3}$

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

|    | $x$        | $y$          | $z$         | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|----|------------|--------------|-------------|----------------------------------|
| O2 | 1.0621 (4) | 0.51786 (13) | 0.89866 (9) | 0.0431 (4)                       |

|     |            |              |              |            |
|-----|------------|--------------|--------------|------------|
| F1  | 1.4484 (3) | 0.55381 (14) | 0.59940 (9)  | 0.0624 (5) |
| N1  | 0.7010 (4) | 0.21726 (17) | 0.71693 (12) | 0.0388 (5) |
| C4  | 1.0013 (4) | 0.38262 (17) | 0.73456 (12) | 0.0295 (5) |
| C7  | 0.9558 (5) | 0.2922 (2)   | 0.59302 (14) | 0.0410 (6) |
| H7  | 0.8767     | 0.2328       | 0.5556       | 0.049*     |
| O1  | 0.8085 (4) | 0.36747 (14) | 0.95496 (9)  | 0.0544 (5) |
| H1  | 0.8432     | 0.4097       | 0.9970       | 0.082*     |
| C5  | 0.8845 (5) | 0.29428 (17) | 0.67573 (14) | 0.0322 (5) |
| C3  | 0.9254 (5) | 0.41935 (19) | 0.89110 (13) | 0.0341 (5) |
| C6  | 1.1989 (4) | 0.47175 (18) | 0.70871 (13) | 0.0333 (5) |
| H6  | 1.2837     | 0.5308       | 0.7456       | 0.040*     |
| C2  | 0.8776 (5) | 0.35426 (18) | 0.81256 (13) | 0.0320 (5) |
| C1  | 0.6979 (5) | 0.25319 (19) | 0.79748 (14) | 0.0367 (5) |
| H1A | 0.5906     | 0.2154       | 0.8368       | 0.044*     |
| C9  | 1.2604 (5) | 0.4674 (2)   | 0.62638 (14) | 0.0387 (5) |
| C8  | 1.1466 (5) | 0.3805 (2)   | 0.56824 (14) | 0.0441 (6) |
| H8  | 1.1984     | 0.3819       | 0.5136       | 0.053*     |
| H1B | 0.599 (5)  | 0.154 (2)    | 0.6945 (16)  | 0.052 (7)* |

*Atomic displacement parameters ( $\text{\AA}^2$ )*

|    | $U^{11}$    | $U^{22}$    | $U^{33}$    | $U^{12}$     | $U^{13}$    | $U^{23}$     |
|----|-------------|-------------|-------------|--------------|-------------|--------------|
| O2 | 0.0677 (11) | 0.0344 (9)  | 0.0267 (8)  | -0.0101 (8)  | 0.0033 (8)  | -0.0006 (6)  |
| F1 | 0.0725 (10) | 0.0703 (10) | 0.0478 (9)  | -0.0232 (8)  | 0.0220 (8)  | 0.0004 (7)   |
| N1 | 0.0464 (11) | 0.0336 (10) | 0.0360 (11) | -0.0070 (9)  | 0.0027 (9)  | -0.0069 (8)  |
| C4 | 0.0299 (10) | 0.0297 (11) | 0.0280 (11) | 0.0059 (8)   | -0.0011 (9) | -0.0021 (9)  |
| C7 | 0.0449 (13) | 0.0454 (14) | 0.0320 (13) | 0.0019 (11)  | 0.0012 (10) | -0.0135 (10) |
| O1 | 0.0903 (14) | 0.0477 (10) | 0.0262 (9)  | -0.0222 (9)  | 0.0116 (9)  | -0.0019 (7)  |
| C5 | 0.0328 (11) | 0.0304 (11) | 0.0326 (12) | 0.0038 (9)   | 0.0005 (9)  | -0.0034 (9)  |
| C3 | 0.0436 (12) | 0.0324 (11) | 0.0253 (11) | 0.0029 (10)  | 0.0004 (9)  | 0.0039 (9)   |
| C6 | 0.0339 (11) | 0.0332 (11) | 0.0318 (12) | 0.0017 (9)   | -0.0002 (9) | -0.0039 (9)  |
| C2 | 0.0383 (11) | 0.0298 (11) | 0.0270 (11) | 0.0016 (9)   | 0.0006 (9)  | -0.0002 (9)  |
| C1 | 0.0449 (12) | 0.0331 (11) | 0.0322 (12) | 0.0008 (10)  | 0.0046 (10) | 0.0006 (9)   |
| C9 | 0.0378 (12) | 0.0425 (13) | 0.0372 (13) | -0.0015 (10) | 0.0099 (10) | 0.0018 (10)  |
| C8 | 0.0489 (14) | 0.0565 (15) | 0.0281 (12) | 0.0038 (12)  | 0.0092 (11) | -0.0048 (11) |

*Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )*

|        |           |        |           |
|--------|-----------|--------|-----------|
| O2—C3  | 1.246 (2) | C7—H7  | 0.9300    |
| F1—C9  | 1.369 (2) | O1—C3  | 1.328 (2) |
| N1—C1  | 1.351 (3) | O1—H1  | 0.8200    |
| N1—C5  | 1.394 (3) | C3—C2  | 1.444 (3) |
| N1—H1B | 0.89 (3)  | C6—C9  | 1.378 (3) |
| C4—C5  | 1.414 (3) | C6—H6  | 0.9300    |
| C4—C6  | 1.411 (3) | C2—C1  | 1.377 (3) |
| C4—C2  | 1.454 (3) | C1—H1A | 0.9300    |
| C7—C8  | 1.378 (3) | C9—C8  | 1.392 (3) |
| C7—C5  | 1.397 (3) | C8—H8  | 0.9300    |

## supplementary materials

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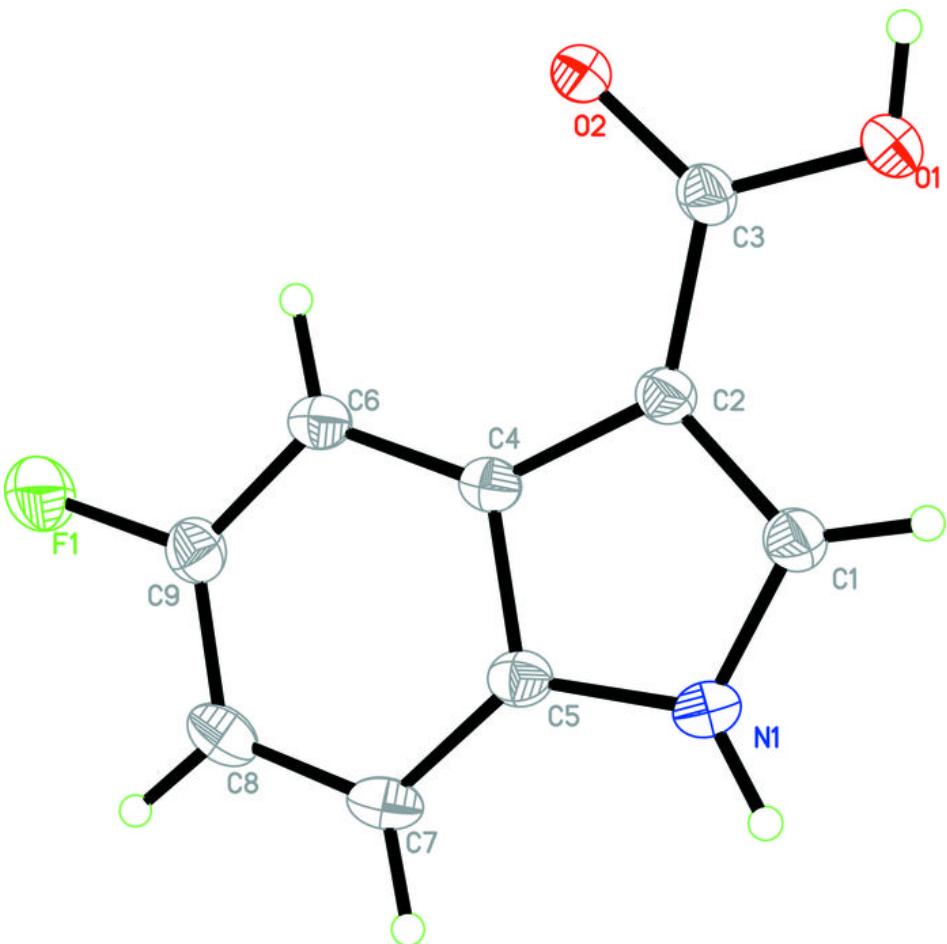
|           |             |           |             |
|-----------|-------------|-----------|-------------|
| C1—N1—C5  | 110.02 (18) | C9—C6—C4  | 116.95 (19) |
| C1—N1—H1B | 123.9 (16)  | C9—C6—H6  | 121.5       |
| C5—N1—H1B | 126.1 (16)  | C4—C6—H6  | 121.5       |
| C5—C4—C6  | 118.62 (19) | C1—C2—C3  | 125.57 (19) |
| C5—C4—C2  | 106.17 (18) | C1—C2—C4  | 106.92 (18) |
| C6—C4—C2  | 135.20 (19) | C3—C2—C4  | 127.49 (19) |
| C8—C7—C5  | 118.3 (2)   | N1—C1—C2  | 109.7 (2)   |
| C8—C7—H7  | 120.9       | N1—C1—H1A | 125.1       |
| C5—C7—H7  | 120.9       | C2—C1—H1A | 125.1       |
| C3—O1—H1  | 109.5       | F1—C9—C6  | 118.10 (19) |
| C7—C5—N1  | 130.24 (19) | F1—C9—C8  | 117.23 (19) |
| C7—C5—C4  | 122.58 (19) | C6—C9—C8  | 124.7 (2)   |
| N1—C5—C4  | 107.18 (18) | C7—C8—C9  | 118.9 (2)   |
| O2—C3—O1  | 122.08 (19) | C7—C8—H8  | 120.6       |
| O2—C3—C2  | 122.83 (19) | C9—C8—H8  | 120.6       |
| O1—C3—C2  | 115.09 (19) |           |             |

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

| $D\cdots H$                            | $D—H$    | $H\cdots A$ | $D\cdots A$ | $D—H\cdots A$ |
|--|----------|-------------|-------------|---------------|
| O1—H1B <sup>i</sup> —O2 <sup>i</sup>   | 0.82     | 1.86        | 2.669 (2)   | 171           |
| N1—H1B <sup>ii</sup> —O2 <sup>ii</sup> | 0.89 (3) | 2.18 (3)    | 3.026 (2)   | 159 (2)       |

Symmetry codes: (i)  $-x+2, -y+1, -z+2$ ; (ii)  $-x+3/2, y-1/2, -z+3/2$ .

Fig. 1



## supplementary materials

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Fig. 2

